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Convergent synthesis of (+)-aspergillide B via a highly diastereoselective oxocarbenium allylation

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Marine-derived natural products continue to provide medici-nally relevant lead compounds for a variety of therapeutic areas.^{[1](#page-2-0)} One such family of biologically active compounds is the aspergillides as they were recently isolated in 2008 from the marine fungus Aspergillus ostianus strain 01F313 cultured within a bro-mine-modified medium by Kusumi and co-workers.^{[2](#page-2-0)} The aspergillides A (1) , B (2) , and C (3) are 14-membered lactones that contain either an α or a β -C-glycoside subunit and a *trans*-olefin as part of the macrocyclic structure. In addition to their intriguing structures, natural products 1–3 have shown cytotoxic properties against the L1210 murine leukemia cell line with LD_{50} values ranging from 2.0-71 µg/ml. However, the initial disclosure of Kusumi assigned the structures of aspergillides A and B incorrectly. The chemical synthesis of the presumed aspergillide A by Uenishi provided aspergillide B and shed light on a structural discrepancy between these two natural products.^{[3](#page-2-0)} Subsequent to this Letter, Kusumi and Ooi published structural revisions of both aspergillides A and B by means of X-ray crystallography.[4](#page-2-0) The corrected structures of 1–3 are shown in Figure 1.

Based on the limited biological data and unique structural features of compounds 1–3, there has been great synthetic interest in these natural products.⁵⁻⁷ Herein, we disclose our synthesis of (+)-aspergillide B based on a stereoselective oxocarbenium allylation to forge the α -C-glycoside subunit.

Our synthetic blueprint to *ent*-2 was envisioned to feature a Yamaguchi macrolactonization for the completion of the 14-mem-bered ring as shown in [Figure 2](#page-1-0).^{[8](#page-2-0)} A diastereoselective crossmetathesis was reserved for the convergent coupling of two fragments and the α -C-glycoside segment would arise from a stereoselective oxocarbenium allylation process followed by a Ru-catalyzed terminal olefin isomerization protocol, vide infra.

Our synthetic outline to *ent*-2 required the synthesis of the chiral alkenol 10 as delineated in [Scheme 1.](#page-1-0) Hence, a copper-catalyzed Grignard addition of 3-butenyl magnesium bromide to (R)-propyl-

ene oxide (9) readily provided the desired secondary alkenol in 88% yield. Protection of the free hydroxyl group was achieved with the standard silylating reagents (TBSCl and imidazole) in DMF and furnished the TBS-protected alkenol 10 in 94% yield.

With the chiral alkenol segment 10 readily in hand and in multi-gram quantities, we next focused our effort on the completion of the α -C-glycoside portion of **ent-2** as delineated in [Schemes 2 and](#page-1-0) [3](#page-1-0). Thus, treatment of the TBS-protected aldehyde 11 with the chiral pinane Z-MEM-acetal crotyl borane reagent provided the MEMprotected diol 12 with a dr of >95:5 for the syn diastereomer and $94%$ ee with a 75% yield.⁹ An ensuing ester formation of 12 with acryloyl chloride in the presence of Et_3N and DMAP afforded the acrylate ester 13 in 85% yield. With 13 in hand, we proceeded to form the α , β -unsaturated lactone via olefin metathesis. Hence, treatment of 13 with Grubbs' second generation carbene catalyst 14 under high dilution provided lactone 15 in a very good yield of 96%[.10](#page-2-0) Subsequent reduction of the olefin resident in 15 with $H₂$ and Pd/C under atmospheric pressure furnished lactone 8. With 8 in hand, partial reduction of the lactone moiety with DIBAL afforded the intermediate lactol and final capping of the hemiacetal with Ac_2O provided the oxocarbenium precursor 16 in 79% yield over two steps from 8.

With acetate precursor 16 in hand, we proceeded to forge the α -C-glycoside subunit 7 by means of a stereoselective oxocarbenium allylation. Based on our previous synthetic ventures, we were hopeful that this approach would stereoselectively deliver the targeted a-C-glycoside with high levels of dr and in good to excellent

Figure 1. Corrected structures of aspergillides $A(1)$, $B(2)$, and $C(3)$.

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Figure 2. Retrosynthetic analysis of ent-2.

Scheme 1. Synthesis of intermediate 10. Reagents and conditions: (a) 3-butenyl magnesium bromide (2.0 equiv), CuCN (10 mol %), THF, -78 °C, 12 h, 88%; (b) TBSCl (1.5 equiv), imidazole (3.0 equiv), DMF, 0 \degree C to rt, 12 h, 94%.

Scheme 2. Synthesis of intermediate 16. Reagents and conditions: (a) BF_3/OE_2 (1.3 equiv), $(-)$ -Ipc₂BCH₂CH=CHOMEM (1.1 equiv), THF, -78 °C, 3 h, 75%; (b) acryloyl chloride (2.0 equiv), NEt₃ (2.2 equiv), DMAP (5 mol %), CH₂Cl₂, 0 °C to rt, 12 h, 85%; (c) 14 (15 mol %), toluene, 80 °C, 24 h, 96%; (d) Pd/C (10%), EtOH, rt, 1.5 h, 99%; (e) DIBAL-H (2.5 equiv), CH_2Cl_2 , -78 °C, 4 h, 99%; (f) Ac₂O (2.5 equiv), pyr (3.0 equiv), DMAP (1.0 equiv), CH₂Cl₂, rt, 12 h, 80%.

yield. 11 Hence, treatment of acetate **16** with BF3 \cdot OEt $_2$ at -78 $^{\circ}$ C led to the formation of an oxocarbenium cation which has two possible reactive conformers 17 and 18 as shown in Scheme 3. Conformer 17 should place the C5 substituent in the pseudo axial

Scheme 3. Synthesis of intermediate 7. Reagents and conditions: (a) BF_3/OE_2 (1.5 equiv), allylTMS (2.5 equiv), CH_2Cl_2 , -78 °C, 1 h, 85%.

position while the MEM acetal at C4 would be in the pseudo equatorial position. Contrary to 17, conformer 18 would place the C5 side chain into the pseudo equatorial orientation while the MEM acetal is axial at C4. Based on our previous observations and in agreement with Woerpel's reports, we predicted that conformer 18 would be the reactive one and the allyITMS nucleophile would stereoselectively approach 18 from the axial position leading to the chair-like transition state via the stereoelectronic effect.¹¹⁻¹³ As expected, the α -C-glycoside 7 was isolated with a very high dr $($ >20:1) as a single stereoisomer in 85% yield.^{[14](#page-2-0)}

With the completion of the α -C-glycoside, an olefin isomerization of 7 prior to the convergent cross-metathesis was required. Inspired by Hanessian's Letter of terminal olefin isomerization

Scheme 4. Synthesis of intermediate 4. Reagents and conditions: (a) 14 (10 mol %), MeOH, 60 °C, 6 h, 65%; (b) 10 (5.0 equiv), 14 (10 mol %), CH₂Cl₂, 45 °C, 8 h, 67%; (c) TBAF (3 equiv), THF, $0 \,^{\circ}\text{C}$ to rt, 3 h, 65%; (d) TEMPO (10 mol %), PhI(OAc)₂ (1.1 equiv), CH_2Cl_2 , rt, 12 h, 70%; (e) NaClO₂ (6 equiv), NaH₂PO₄ (6 equiv), 2methyl-2-butene (8 equiv), t-BuOH/H₂O (3:1), rt, 95%.

Scheme 5. Synthesis of (+)-aspergillide B (ent-2). Reagents and conditions: (a) Et₃N/3HF (30 equiv), CH₃CN, 45 °C, 6 h, 63%; (b) 22 (3.9 equiv), NEt₃ (6.0 equiv), DMAP (30 equiv), toluene, 100 °C, 22 h, 75%; (c) LiBF₄ (30 equiv), CH₃CN/H₂O, 72 °C, 5 h, 85%.

utilizing Ru catalyst 14 in MeOH, we sought to utilize this procedure prior to the convergence of 6 and 10 as shown in [Scheme](#page-1-0) [4](#page-1-0).¹⁵ Thus, treatment of **7** with catalyst **14** in MeOH readily afforded the isomerized olefin product 6 in 65% yield and as an $8/1$: E/Z mixture. Subsequently, an E selective cross-metathesis of 6 and 10 with catalyst 14 was accomplished and provided compound 5 in 67% yield with an E/Z ratio of >20:1.¹⁶ Selective deprotection of the primary TBS group with TBAF in the presence of a secondary one furnished the free primary alcohol 19 in 65% yield. An ensuing oxidation of the primary alcohol resident in 19 with TEMPO and PhI(OAc)₂ readily afforded aldehyde 20 in 70% yield.¹⁷ Further oxidation of the aldehyde moiety to the carboxylic acid was accomplished in 95% yield under the Lindgren-Kraus-Pinnick conditions to provide acid 4, and set the final stage of macrocyclization to provide *ent*-2.¹⁸

With the TBS-protected carboxylic acid 4 in hand, we focused our attention to the final macrocyclization and deprotection to afford ent-2 as highlighted in Scheme 5. With this in mind, deprotection of the TBS group was accomplished with Et₃N·3HF and provided the free hydroxy-carboxylic acid compound 21 in 63% yield. An ensuing Yamaguchi macrocyclization of 21 under the reported standard conditions with 22 , Et₃N, and DMAP provided the MEMprotected macrolactone 23 in 75% yield.⁸ Final deprotection of 23 with LiBF₄ furnished **ent-2** in 85% yield. The spectral data (¹H NMR, 500 MHz; ¹³C NMR, 125 MHz) and HRMS data of synthetic **ent-2** were in agreement with the natural sample.^{2,19} However, the optical rotation ($[\alpha]_{D}^{\text{rt}}$ +81.1, c = 0.0041 g/ml MeOH) confirmed that **ent-2** is the enantiomer of the natural aspergillide B.

In conclusion, the synthesis of (+)-aspergillide B has been accomplished by means of a highly diastereoselective oxocarbenium allylation and a Ru-catalyzed terminal olefin isomerization followed by a cross-metathesis and final Yamaguchi macrolactonization. The late stage convergence allows for the synthesis of a variety of analogues to examine the bioactivity of structurally diverse 'aspergillide-like' compounds against a collection of tumor cell lines and further develop the SAR of the aspergillide family of natural products.

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References and notes

- 1. Bugni, T. S.; Ireland, C. M. Nat. Prod. Rep. 2004, 21, 143.
- 2. Kito, K.; Ookura, R.; Yoshida, S.; Namikoshi, M.; Ooi, T.; Kusumi, T. Org. Lett. 2008, 10, 225.
- 3. Hande, S. M.; Uenishi, J. Tetrahedron Lett. 2009, 50, 189.
- 4. Ookura, R.; Kito, K.; Saito, Y.; Kusumi, T.; Ooi, T. Chem. Lett. 2009, 38, 384.
- 5. Syntheses of aspergillide A: (a) Nagasawa, T.; Kuwahara, S. Tetrahedron Lett. 2010, 51, 875; (b) Diaz-Oltra, S.; Angulo-Pachón, C.; Murga, J.; Carda, M.; Marco, J. A. J. Org. Chem. 2010, 75, 1775; (c) Fuwa, H.; Yamaguchi, H.; Sasaki, M. Org. Lett. 2010, 12, 1848.
- 6. Syntheses of aspergillide B: (a) Diaz-Oltra, S.; Angulo- Pachón, C.; Kneeteman, M. N.; Murga, J.; Carda, M.; Marco, J. A. Tetrahedron Lett. 2009, 50, 3783; (b) Nagasawa, T.; Kuwahara, S. Biosci. Biotechnol. Biochem. 2009, 73, 1893; (c) Liu, J.; Xu, K.; He, J.; Zhang, L.; Pan, X.; She, X. J. Org. Chem. 2009, 74, 5063.
- 7. Syntheses of aspergillide C: (a) Nagasawa, T.; Kuwahara, S. Org. Lett. 2009, 11, 761; (b) Panarese, J. D.; Waters, S. P. Org. Lett. 2009, 11, 5086.
- 8. Yamaguchi, M.; Katsuki, T.; Saeki, H.; Hirata, K.; Inanaga, J. Bull. Chem. Soc. Jpn. 1979, 52, 1989.
- 9. (a) Brown, H. C.; Jadhav, P. K.; Bhat, K. S. J. Am. Chem. Soc. 1988, 110, 1535; (b) Smith, A. L.; Hwang, C. K.; Pitsinos, E.; Scarlato, G. R.; Nicolaou, K. C. J. Am. Chem. Soc 1992, 114, 3134; (c) Ramachandran, P. V.; Chandra, J. S.; Reddy, M. V. R. J. Org. Chem. 2002, 67, 7547.
- 10. For a very recent review, see: (a) Vougioukalakis, G. C.; Grubbs, R. H. Chem. Rev. 2010, 110, 1746; (b) Scholl, M.; Ding, S.; Lee, C. W.; Grubbs, R. H. Org. Lett. 1999, 1, 953.
- 11. (a) Jennings, M. P.; Clemens, R. T. Tetrahedron Lett. 2005, 46, 2021; (b) Ding, F.; Jennings, M. P. Org. Lett. 2005, 7, 2321; (c) Clemens, R. T.; Jennings, M. P. Chem. Commun. 2006, 2720; (d) Sawant, K. B.; Jennings, M. P. J. Org. Chem. 2006, 71, 7911; (e) Sawant, K. B.; Ding, F.; Jennings, M. P. Tetrahedron Lett. 2006, 47, 939; (f) Sawant, K. B.; Ding, F.; Jennings, M. P. Tetrahedron Lett. 2007, 48, 5177; (g) Ding, F.; Jennings, M. P. J. Org. Chem. 2008, 73, 5965; (h) Carrick, J. D.; Jennings, M. P. Org. Lett. 2009, 11, 769.
- 12. (a) Romero, J. A. C.; Tabacco, S. A.; Woerpel, K. A. J. Am. Chem. Soc. 2000, 122, 168; (b) Ayala, L.; Lucero, C. G.; Romero, J. A. C.; Tabacco, S. A.; Woerpel, K. A. J. Am. Chem. Soc. 2003, 125, 15521.
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- 13. Corey, E. J.; Sneen, R. A. J. *Am. Chem. Soc.* **1956**, 78, 6269.
14. *Data for 7*: ¹H NMR (500 MHz, CDCl₃) δ 5.8 (m, 1H), 5.0 (m, 2H), 4.7 (dd, J = 15.9, 6.8 Hz, 2H), 4.0 (m, 1H), 3.7 (m, 6H), 3.5 (m, 2H), 3.3 (s, 3H), 2.2 (m, 2H), 1.9 (m, 1H), 1.8 (m, 2H), 1.7 (m, 2H), 1.3 (m, 1H), 0.09 (s, 9H), 0.03 (s, 6H). 13C NMR $(125 \text{ MHz}, \text{CDCl}_3) \delta 135.2, 116.5, 94.1, 72.9, 71.7, 70.3, 69.2, 67.0, 59.8, 58.9,$ 38.4 , 29.7, 27.7, 25.9, 24.2, 18.3, -5.3. IR (CH₂Cl₂): 3100, 2932, 2874, 2359, 1640, 1463, 1250, 1092, 1046, 843, 780 cm⁻¹. $R_f = 0.57$, 30% EtOAc in hexanes. $[\alpha]_D^{24}$ +21.4 (c 0.013 g/mL, CH₂Cl₂). HRMS (EI) calcd for C₂₀H₄₀O₅Si [M]⁺: 389.2723; found: 334.2721.
- 15. Hanessian, S.; Giroux, S.; Larsson, A. Org. Lett. 2006, 8, 5481.
- 16. Data for 5: ¹H NMR (500 MHz, CDCl₃) δ 5.5 (m, 2H), 4.7 (dd, J = 15.9, 6.8 Hz, 2H), 4.2 (m, 1H), 3.9 (m, 1H), 3.7 (m, 6H), 3.5 (m, 2H), 3.4 (s, 3H), 1.9 (m, 4H), 1.7 (m,
4H), 1.4 (m, 4H), 1.1 (d, J = 6.1 Hz, 3H), 0.9 (s, 18H), 0.04 (s, 12H). ¹³C NMR $(125 \text{ MHz}, \text{CDCl}_3)$ δ 132.5, 129.7, 94.1, 72.7, 71.7, 70.8, 69.7, 68.4, 67.0, 59.7, 59.0, 39.2, 32.4, 25.9, 25.8, 25.3, 23.9, 23.7, 18.3, 18.1, –4.7. IR (CH₂Cl₂): 2930,
2858, 2356, 1466, 1362, 1260, 1089, 1039, 830, 766 cm⁻¹. R_f = 0.80, 30% EtOAc in hexanes. $[\alpha]_D^{24}$ +7.2 (c 0.013 g/mL, CH₂Cl₂). HRMS (EI) calcd for C₃₀H₆₂O₆Si₂ [M]⁺: 575.4163; found: 575.4149.
- 17. De Mico, A.; Margarita, R.; Parlanti, L.; Vescovi, A.; Piancatelli, G. J. Org. Chem. 1997, 62, 6974.
- 18. (a) Lindgren, B. O.; Nilsson, T. Acta Chem. Scand. 1973, 27, 888; (b) Kraus, G. A.; Taschner, M. J. J. Org. Chem. 1980, 45, 1175; (c) Kraus, G. A.; Roth, B. J. Org. Chem. 1980, 45, 4825; (d) Pinnick, H. W.; Childers, W. E.; Bal, B. S. Tetrahedron 1981, 37, 2091.
- 19. Data for **ent-2:** ¹H NMR (500 MHz, benzene-d) δ 6.1 (dddd, J = 15.6, 10.8, 4.8 1.6 Hz, 1H), 5.3 (dd, J = 15.6, 4.4 Hz, 1H), 5.0 (m, 1H), 4.3 (m, 1H), 4.0 (d, $J = 11.2$ Hz, 1H), 3.2 (br s, 1H), 2.7 (dd, $J = 13.6$, 11.6 Hz, 1H), 2.1 (dd, $J = 13.6$, 2.0 Hz, 1H), 2.0 (dddd, J = 13.6, 10.8, 4.8, 2.4 Hz, 1H), 1.8 (br s, 1H), 1.7-1.8 (m, 2H), 1.5–1.6 (3H, m), 1.3–1.4 (m, 3H), 1.0 (d, J = 6.4 Hz, 3H), 0.9 (m, 1H). 13C NMR (125 MHz, benzene-d) δ 169.8, 138.2, 129.1, 71.6, 69.9, 69.6, 67.3, 39.9, 32.1, 30.8, 27.8, 25.3, 22.6, 19.2. IR (CH₂Cl₂): 3381, 2926, 1717, 1267, 187, 1387, 1086, 1017, 965 cm⁻¹. $R_f = 0.33$, 60% EtOAc in hexanes. [$\alpha_{\text{lb}}^{2/4}$ +81.1 (*c* 0.0041 g/mL, MeOH). HRMS (EI) calcd for $C_{14}H_{22}O_4$ [M]⁺: 2547.1518; found 254.1513.